### **Cyanide Poisoning**

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## Sources

- Cyanide (hydrocyanic acid, prussic acid)
- Sodium nitroprusside
- Fire victims
- Trace are produced indigenously from Vitamin B12 metabolism
- Amygdalins, which are hydrolyzed to hydrogen cyanide is present in the seeds of apple, peach, plum, apricot, cherry, and bitter almond.
- Industrial chemicals;
- Electroplating, electro-polishing, extraction of gold and silver from ores, plastic manufacture, fumigant

## Mechanism of Toxicity

- Histotoxic anoxia
- It results
  - From binding of CN with the Ferric ion on the a-a<sup>3</sup>
    complex within the Cytochrome complex
  - The tissue will not be able to utilize Oxygen
  - Anaerobic respiration
  - No production of ATP
  - Oxygen will remain in the venous blood

## Toxicity

- Metabolism:
  - Rhodanese enzyme → thiocyanate → renal excretion (major pathway)
  - Cobalamine + cyanide  $\rightarrow$  Cyanocobalamin
  - Excreted via breath and sweat

# Signs and Symptoms cyanide poisoning

- Major organs affected are CNS and cardiovascular system
  - Weakness
  - Dizziness
  - Headache
  - Nausea and vomiting
  - Tachycardia
  - Flushing

### Treatment

- ABC
- Decontamination
- Antidote
- Continuous care

- Antidote :
  - Amyl nitrite inhalation , 0.3 ml
  - Sodium nitrite, 300 mg in 10 ml
  - Sodium thiosulfate, 12.5 g in 50 ml

• Hb-Fe<sup>+2</sup> + NO<sub>2</sub>  $\rightarrow$  Hb-Fe<sup>+3</sup> + NO

- Hb-Fe <sup>+3</sup> + CN-cytochrome-Fe<sup>+3</sup> → CN-HB-Fe<sup>+3</sup>
  + cytochrome-Fe<sup>+3</sup>
- CN-Hb-Fe<sup>+3</sup> +Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> $\rightarrow$ CN-S+Na<sub>2</sub>SO<sub>3</sub>+Hb-Fe<sup>+3</sup>